ACS Catalysis

Correction

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Correction to "Dendrimer Encapsulated Copper Cluster as a Chemoselective and Regenerable Hydrogenation Catalyst"

Prasenjit Maity, Seiji Yamazoe, and Tatsuya Tsukuda*

ACS Catalysis **2013**, *3*, 182−185

On page 182, ref 9 (Vilar-Vidal, N.; Rivas, J.; López-Quintela, M. A. *ACS Catal.* **2012**, *2*, 1693) was cited improperly in the following two sentences:

- For example, Cu nanoparticles (NPs) stabilized by organic polymers or immobilized on solid supports are easily oxidized into CuO or Cu_2O NPs under ambient conditions[3−6] with a few exceptions where they are transformed into Cu^{2+} ions.[7−9]
- Indeed, Cu NPs (>10 nm) have been used as catalysts for reduction of a dye by hydrazine[9] and aromatic nitro compounds[14] and azides[15] by ammonium formate, hydrogenation of nitrophenol[16] and CO_2,[17−19] and in the water-gas shift reaction.[20]

Those two sentences should be corrected, respectively, as follows.

- For example, Cu nanoparticles (NPs) stabilized by organic polymers or immobilized on solid supports are easily oxidized into Cu_xO (x = 1, 2) NPs[3−6] or Cu^{2+} ions[7,8] under ambient conditions, with the exception of where Cu_n (n = 5, 13, 20) clusters stabilized by tetrabutylammonium nitrate are stable against oxidation.[9]
- Indeed, Cu NPs have been used as catalysts for reduction of aromatic nitro compounds[14] and azides[15] by ammonium formate, hydrogenation of nitrophenol[16] and CO_2[17−19] and in the water-gas shift reaction.[20]

We appreciate Dr. Vilar-Vidal's pointing out the erroneous citations.

AUTHOR INFORMATION

Notes

The authors declare no competing financial interest.

Published: March 1, 2013

dx.doi.org/10.1021/cs400117x | *ACS Catal.* 2013, 3, 554−554